

## Answers

### Part 1: The Orion Nebula

#### Observation table

| Picture name                       | Colours                         | What makes the colours?   |
|------------------------------------|---------------------------------|---|
| The Orion nebula                   | Red<br>Green<br>Yellow<br>Black | Atomic hydrogen<br>Molecular oxygen<br>Hydrogen and oxygen mixed<br>Clouds of dust, Interstellar Medium (ISM) |
| The Horsehead nebula               | Black<br>Red<br>Blue            | Clouds of dust, ISM<br>Atomic hydrogen<br>Reflected starlight   |
| Reflection nebula in Orion's sword | Blue<br>Red<br>Black            | Reflected starlight<br>Atomic hydrogen<br>Clouds of dust, ISM   |

#### Questions

- Atomic hydrogen
- Atomic hydrogen is the simplest substance and the starting point for all other chemical elements, so is very common in the Universe.
- A large cloud of dust.
- The light is too faint to be seen by the naked eye.

### Part 2: The Interstellar Medium (ISM)

|                                    |  |
|------------------------------------|--|
| The Orion nebula                   | Atomic hydrogen<br>Molecular oxygen<br>Hydrogen and oxygen mixed<br>Clouds of dust, Interstellar Medium (ISM)                            |
| The Horsehead nebula               | Clouds of dust, ISM<br>Atomic hydrogen<br>Reflected starlight  |
| Reflection nebula in Orion's sword | Reflected starlight<br>Atomic hydrogen<br>Clouds of dust, ISM  |
| Picture 1                          | Yellow: mixture of atomic hydrogen and molecular oxygen<br>Black: dust, ISM<br>Red: Atomic hydrogen<br>Blue: Starlight reflected by dust |
| Picture 2                          | In the regions where there are stars.  |
| Picture 3                          | In the dark clouds.  |
| Picture 4                          | In the dark clouds.  |

The ISM contains complex molecules and simple molecules. Chemistry must be happening in the ISM, ie new bonds being formed between atoms. The stuff in the ISM comes from stars and new bond formation.

Other questions are to be answered by student research.

## Acknowledgements

V. Kind, *Contemporary chemistry for schools and colleges*. London: Royal Society of Chemistry, 2004.



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