

Nuclear decay equations 3

Education in Chemistry

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This worksheet accompanies the above article 'The hunt is on'.

Complete the equations by adding in the type of decay, the isotope undergoing decay and/or the products. Each equation has one or more parts missing.

	Type of decay					
1		${}_{19}^{40}\text{K}$	\longrightarrow	${}_{20}^{40}\text{Ca}$	+	
2			\longrightarrow	${}_{84}^{206}\text{Po}$	+	${}_{2}^{4}\text{He}$
3	α	${}_{90}^{230}\text{Th}$			+	${}_{2}^{4}\text{He}$
4				${}_{91}^{234}\text{Pa}$	+	${}_{-1}^{0}\text{e}$
5	α	${}_{4}^{10}\text{Be}$				
6		${}_{4}^{8}\text{Be}$		${}_{5}^{8}\text{B}$	+	${}_{-1}^{0}\text{e}$
7	β	${}_{34}^{79}\text{Se}$				
8	α			${}_{82}^{206}\text{Pb}$	+	
9		${}_{36}^{85}\text{Kr}$		${}_{37}^{85}\text{Rb}$	+	
10	β	${}_{38}^{90}\text{Sr}$				
11				${}_{86}^{234}\text{Rn}$	+	${}_{2}^{4}\text{He}$
12	α			${}_{92}^{234}\text{U}$	+	${}_{2}^{4}\text{He}$
13	β	${}_{38}^{125}\text{Sb}$				
14	α	${}_{96}^{242}\text{Cm}$				
15	α	${}_{83}^{209}\text{Bi}$			+	${}_{2}^{4}\text{He}$