

Expressing uncertainty in data

Education in Chemistry July 2020 rsc.li/2Xfd61c

Practise handling significant figures, decimal places and uncertainty in measurements.

1. Rounding

a. Answer the questions using the number lines.



b. Fill in the blanks on the number lines and answer the questions.



c. Circle the number that is the nearest 1000, 100 and 10 to the number mentioned in each question and indicate whether it is rounded up or down.

i	2527
	2021

••	
	1168
	4400

	nearest	nearest	nearest		nea	rest	nearest	nearest
	1000	100	10		10	00	100	10
	3000	2600	2530		50	00	4500	4460
	2000	2500	2520		40	00	4400	4450
up or				up	or			
down?				dow	/n?			

d. Circle the number that is the nearest integer, 0.1 and 0.01 and indicate whether it is rounded up or down.

iii. 2.527

iv. 4.458

	nearest	nearest	nearest
	integer	0.1	0.01
	3	2.6	2.53
	2	2.5	2.52
up or			
down?			

	nearest	nearest	nearest
	integer	0.1	0.01
	5	4.5	4.46
	4	4.4	4.45
up or			
down?			

e. Insert the missing numbers so that the table looks like the ones above. Then, circle the number that is the nearest integer, 0.1 and 0.01 and indicate whether it is rounded up or down.

i. 4.097

	nearest	nearest	nearest
	integer	0.1	0.01
		4.1	
			4.09
up or			
down?			

ii. 10.893

	nearest	nearest	nearest
	integer	0.1	0.01
		10.9	
			10.89
up or			
down?			

2. Significant figures Circle the correct answer.

a. Round 34.59 to 1 significant figure					
A: 34	B: 30	C: 35	D: 3	E: 34.5	
b. Round 35,683	to 1 significant figure	;			
A: 35683	B: 30000	C: 40000	D: 3	E: 4	
				·	
c. Round 76.984 t	o 3 significant figures	6			
A: 77.0	B: 76.9	C: 76.984	D: 77	E: 76	
d. Round 0.003865 to 1 significant figure					
A: 0	B: 0.003	C: 0.004	D: 0.00387	E: 1	
f. Round 0.003865 to 3 significant figures					
A: 0.00	B: 0.003	C: 0.004	D: 0.00386	E: 0.00387	

3. Chemistry contexts

- a. Calculate the rate of reaction that produces 25.0 cm³ of gas in 5.90 s.
- b. Calculate the rate of reaction that produces 25 cm³ of gas in 5.90 s.
- c. Calculate the rate of reaction that produces 25.0 cm³ of gas in 5.9 s.
- d. Calculate the rate of reaction that produces 25 cm³ of gas in 5.9 s.
- e. Calculate the concentration of a sodium chloride solution when 9.80 g of solid was dissolved in 100 cm³ water.
- f. Calculate the concentration of a sodium chloride solution when 9.8 g of solid was dissolved in 100 cm³ water.
- g. Calculate the concentration of a sodium chloride solution when 0.98 g of solid was dissolved in 10 cm³ water.
- h. Calculate the concentration of a sodium chloride solution 0.98 g of solid was dissolved in 10.0 cm³ water.