## Expressing uncertainty in data

## Education in Chemistry

July 2020
rsc.li/2Xfd61c
Practise handling significant figures, decimal places and uncertainty in measurements.

## 1. Rounding

a. Answer the questions using the number lines.

b. Fill in the blanks on the number lines and answer the questions.
i. Draw an arrow to indicate where 203 is on a number line.
ii. What is 203 to the nearest hundred?

iii. Draw an arrow to indicate where 3.089 is on a number line.
iv. What is 3.089 to 2 decimal places?

c. Circle the number that is the nearest 1000,100 and 10 to the number mentioned in each question and indicate whether it is rounded up or down.

## i. 2527

|  | nearest <br> 1000 | nearest <br> 100 | nearest <br> 10 |
| :--- | :---: | :---: | :---: |
|  | 3000 | 2600 | 2530 |
|  | 2000 | 2500 | 2520 |
| up or <br> down? |  |  |  |

ii. 4458

|  | nearest <br> 1000 | nearest <br> 100 | nearest <br> 10 |
| :--- | :---: | :---: | :---: |
|  | 5000 | 4500 | 4460 |
| $l \mid$ | 4000 | 4400 | 4450 |
| up or <br> down? |  |  |  |

d. Circle the number that is the nearest integer, 0.1 and 0.01 and indicate whether it is rounded up or down.
iii. 2.527

|  | nearest <br> integer | nearest <br> 0.1 | nearest <br> 0.01 |
| :--- | :---: | :---: | :---: |
|  | 3 | 2.6 | 2.53 |
|  | 2 | 2.5 | 2.52 |
| up or <br> down? |  |  |  |

$$
\text { iv. } 4.458
$$

|  | nearest <br> integer | nearest <br> 0.1 | nearest <br> 0.01 |
| :---: | :---: | :---: | :---: |
|  | 5 | 4.5 | 4.46 |
|  | 4 | 4.4 | 4.45 |
| up or <br> down? |  |  |  |

e. Insert the missing numbers so that the table looks like the ones above. Then, circle the number that is the nearest integer, 0.1 and 0.01 and indicate whether it is rounded up or down.
i. 4.097

| nearest <br> integer | nearest <br> 0.1 | nearest <br> 0.01 |  |
| :--- | :---: | :---: | :---: |
|  |  | 4.1 |  |
|  |  |  | 4.09 |
| up or <br> down? |  |  |  |

ii. 10.893

|  | nearest <br> integer | nearest <br> 0.1 | nearest <br> 0.01 |
| :--- | :--- | :---: | :---: |
|  |  | 10.9 |  |
| up or <br> down? |  |  | 10.89 |

2. Significant figures

Circle the correct answer.

| a. Round 34.59 to 1 significant figure | C: 34.5 |  |  |  |
| :--- | :--- | :--- | :--- | :--- |
| A: 34 | B: 30 | C: 35 | D: 3 |  |


| b. Round 35,683 to 1 significant figure |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- |
| A: 35683 | B: 30000 | C: 40000 | D: 3 | E: 4 |

C. Round 76.984 to 3 significant figures

| A: 77.0 | B: 76.9 | C: 76.984 | D: 77 | E: 76 |
| :--- | :--- | :--- | :--- | :--- |


$\left\lvert\,$| d. Round 0.003865 to 1 significant figure | D: 0.00387 | E: 1 |  |
| :--- | :--- | :--- | :--- | :--- |
| A: 0 | B: 0.003 | C: 0.004 |  |
| f. Round 0.003865 to 3 significant figures D: 0.00386 E: 0.00387   <br> A: 0.00 B: 0.003 C: 0.004   |  |  |  |.\right.

## 3. Chemistry contexts

a. Calculate the rate of reaction that produces $25.0 \mathrm{~cm}^{3}$ of gas in 5.90 s .
b. Calculate the rate of reaction that produces $25 \mathrm{~cm}^{3}$ of gas in 5.90 s .
c. Calculate the rate of reaction that produces $25.0 \mathrm{~cm}^{3}$ of gas in 5.9 s .
d. Calculate the rate of reaction that produces $25 \mathrm{~cm}^{3}$ of gas in 5.9 s .
e. Calculate the concentration of a sodium chloride solution when 9.80 g of solid was dissolved in 100 $\mathrm{cm}^{3}$ water.
f. Calculate the concentration of a sodium chloride solution when 9.8 g of solid was dissolved in 100 $\mathrm{cm}^{3}$ water.
g. Calculate the concentration of a sodium chloride solution when 0.98 g of solid was dissolved in 10 $\mathrm{cm}^{3}$ water.
h. Calculate the concentration of a sodium chloride solution 0.98 g of solid was dissolved in $10.0 \mathrm{~cm}^{3}$ water.

