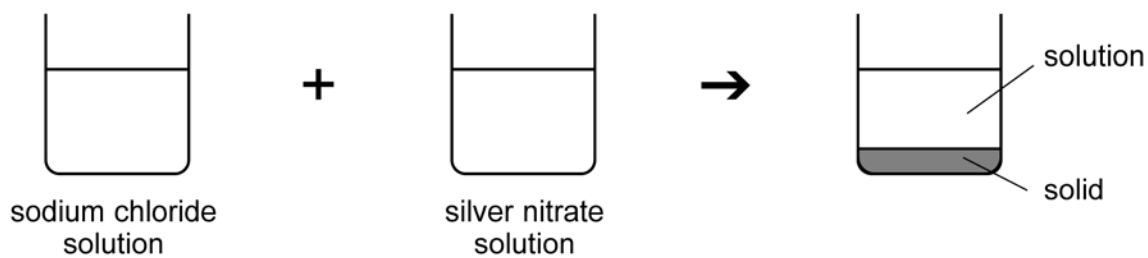


Precipitation reactions

Name



I think that the particles in sodium chloride solution are best represented by diagram
because

I think that the particles in silver nitrate solution are best represented by diagram
because

I think that the particles in the solid, produced when equal amounts of sodium chloride and silver nitrate solutions are mixed, are best represented by diagram

because

I think that the particles in the solution, left behind when equal amounts of sodium chloride and silver nitrate solutions are mixed, are best represented by diagram

because

1. $\text{Pb}^{2+}(\text{aq})$ + $2\text{I}^{-}(\text{aq})$ → (s)
2. $\text{Ba}^{2+}(\text{aq})$ +(aq) → $\text{BaSO}_4(\text{s})$
3.(aq) + $\text{OH}^{-}(\text{aq})$ → $\text{Cu}(\text{OH})_2(\text{s})$

What I have learned from this activity is