## Concentrations

Answers

| Concentration | Mass of NaOH | Volume of water |
| :--- | :--- | :--- |
| $4 \mathrm{~g} \mathrm{dm}^{-3} \mathrm{NaOH}$ |  |  |
| $0.4 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{NaOH}$ |  |  |
| $0.5 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{NaOH}$ |  |  |
| $2.5 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{Na}^{+}$ions |  |  |

The general way that I worked out the answers

Answers to questions set by other students

| Concentration | Mass of ... | Volume of water |
| :--- | :--- | :--- |
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