## gridlocks - can you unlock the grid?

## Some A-level reagents

Before you answer the gridlocks below fill in the table of formula of reagents that you might meet in A-level chemistry:

| $\mathrm{BaCl}_{2} \mathrm{H}_{2}$ |  | $\mathrm{KMnO}_{4}$ |  |
| :---: | :---: | :---: | :---: |
| name | formula | $\mathbf{N a H C O}_{3}$ | name |
| potassium manganate(VII) |  | ethanoic acid |  |
| sodium dichromate(VI) | $\mathrm{Na}_{2} \mathrm{Cr}_{2} \mathrm{O}_{7}$ | silver nitrate | $\mathrm{AgNO}_{3}$ |
| hydrogen peroxide |  | sodium thiosulfate | $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ |
| sodium ethanoate | $\mathrm{CH}_{3} \mathrm{COO}^{-} \mathrm{Na}^{+}$ | ethanedioic acid | $(\mathrm{COOH})_{2}$ |
| barium chloride |  | potassium cyanide | KCN |
| phosphoric acid | $\mathrm{H}_{3} \mathrm{PO}_{4}$ | sodium hydrogen carbonate |  |

## Gridlock 1

Each row, column and $2 \times 2$ box contains information about the first four reagents listed above. Use your problem solving skills and the answers in the table above to fill in the blank boxes.

| name | formula |  |  |
| :--- | :--- | :--- | :--- |
| sodium dichromate (VI) |  |  |  |
|  |  |  |  |

## gridlocks - can you unlock the grid?

Gridlock 2
Each row, column and $2 \times 2$ box contains information about each of the next four reagents in the table.

| name |  | formula |  |
| :---: | :--- | :--- | :--- |
| phosphoric acid |  |  | $\mathrm{BaCl}_{2}$ |
|  |  |  |  |
|  |  |  |  |
| $\mathrm{CH}_{3} \mathrm{COOH}$ |  |  | silver nitrate |

## Gridlock 3

Each row, column and $2 \times 2$ box contains information about each of the last four reagents in the table

| name |  | formula |  |
| :--- | :--- | :--- | :--- |
| ethanedioic acid |  |  |  |
|  |  |  |  |

## grídlocks - can you unlock the grid?

## Some A-level reagents

Before you answer the puzzles below fill in the table of formula of reagents that you might meet in A-level chemistry:

| $\mathrm{BaCl}_{2}$ | $\mathrm{KMnO}_{2} \mathrm{O}_{2}$ |  | $\mathbf{N a H C O}_{3}$ |
| :---: | :---: | :---: | :---: |
| name | formula | name | formula |
| potassium manganate(VII) | $\mathrm{KMnO}_{4}$ | ethanoic acid | $\mathrm{CH} \mathrm{COOH}_{3}$ |
| sodium dichromate(VI) | $\mathrm{Na}_{2} \mathrm{Cr}_{2} \mathrm{O}_{7}$ | silver nitrate | $\mathrm{AgNO}_{3}$ |
| hydrogen peroxide | $\mathrm{H}_{2} \mathrm{O}_{2}$ | sodium thiosulfate | $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ |
| sodium ethanoate | $\mathrm{CH}_{3} \mathrm{COO}^{-} \mathrm{Na}^{+}$ | ethanedioic acid | $(\mathrm{COOH})_{2}$ |
| barium chloride | $\mathrm{BaCl}_{2}$ | potassium cyanide | KCN |
| phosphoric acid | $\mathrm{H}_{3} \mathrm{PO}_{4}$ | sodium hydrogen carbonate | NaHCO |

## Puzzle 1 - answers

Each row, column and $2 \times 2$ box contains information about the first four reagents listed above. Use your problem solving skills and the answers in the table above to fill in the blank boxes.

| name |  | formula |  |
| :---: | :---: | :---: | :---: |
| sodium dichromate(VI) | sodium ethanoate | $\mathrm{H}_{2} \mathrm{O}_{2}$ | $\mathrm{KMnO}_{4}$ |
| hydrogen peroxide | potassium manganate(VII) | $\mathrm{CH}_{3} \mathrm{COO}^{-} \mathrm{Na}^{+}$ | $\mathrm{Na}_{2} \mathrm{Cr}_{2} \mathrm{O}_{7}$ |
| $\mathrm{KMnO}_{4}$ | $\mathrm{H}_{2} \mathrm{O}_{2}$ | sodium dichromate(VI) | sodium ethanoate |
| $\mathrm{CH}_{3} \mathrm{COO}^{-} \mathrm{Na}^{+}$ | $\mathrm{Na}_{2} \mathrm{Cr}_{2} \mathrm{O}_{7}$ | potassium manganate(VII) | hydrogen peroxide |
| formula |  | name |  |

## grídlockS - can you unlock the grid?

Puzzle 2 - answers
Each row, column and $2 \times 2$ box contains information about each of the next four reagents in the table.

| name |  |  |  |
| :---: | :---: | :---: | :---: |
| phosphoric acid | ethanoic acid | formula |  |
| silver nitrate | barium chloride | $\mathrm{AgNO}_{3}$ | $\mathrm{BaCl}_{2}$ |
| $\mathrm{BaCl}_{2}$ |  | $\mathrm{H}_{3} \mathrm{PO}_{4}$ |  |
| $\mathrm{CH}_{3} \mathrm{COOH}$ |  |  |  |

## Puzzle 3 - answers

Each row, column and $2 \times 2$ box contains information about each of the last four reagents in the table.

| name |  | formula |  |
| :---: | :---: | :---: | :---: |
| ethanedioic acid | sodium hydrogen carbonate | KCN | $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ |
| potassium cyanide | sodium thiosulfate | $\mathrm{NaHCO}_{3}$ | $(\mathrm{COOH})_{2}$ |
| $\mathrm{NaHCO}_{3}$ | $(\mathrm{COOH})_{2}$ | sodium thiosulfate | potassium cyanide |
| $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ | KCN | ethanedioic acid | sodium hydrogen carbonate |
| formula |  | name |  |

