

Knowledge check

Subject area: Organic chemistry

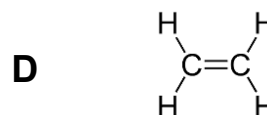
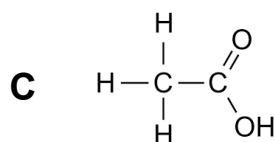
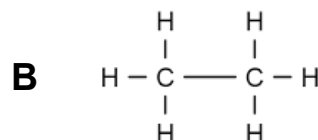
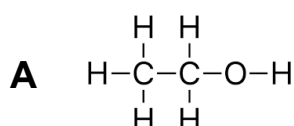
Level: 14–16 years (Foundation)

Topic: Reactions of carboxylic acids

Source: [rsc.li/3o4cneK](https://www.rsc.li/3o4cneK)

1. Each of the organic molecules below belongs to a different functional group.

There is a carboxylic acid, an alcohol, an alkene and an alkane but they are muddled up.



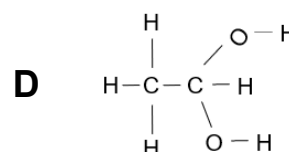
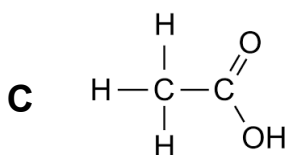
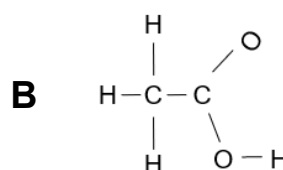
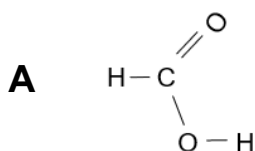
Complete the table by writing each letter (A–D) into the correct box.

Carboxylic acid	Alcohol	Alkene	Alkane

2. George is asked by his teacher to draw out the structure of ethanoic acid.

He gives four possible structures, but only one is correct.

Which one is correct?



3. This question is about naming carboxylic acids.

A student has named the carboxylic acids below, but he has made some mistakes.

How many correct answers can you see?

Give him a mark out of 5 – one mark for each correct answer.

For any incorrect answers, add the correct names.

$\begin{array}{c} \text{H}-\text{C}=\text{O} \\ \\ \text{O}-\text{H} \end{array}$ <p><i>...Methanol...</i></p>	
$\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H}-\text{C}-\text{C}-\text{C}=\text{O} \\ \quad \quad \\ \text{H} \quad \text{H} \quad \text{O}-\text{H} \end{array}$ <p><i>...Pentanoic acid...</i></p>	
$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \\ \quad \quad \quad \\ \text{H}-\text{C}-\text{C}-\text{C}-\text{C}-\text{C}=\text{O} \\ \quad \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \quad \text{O}-\text{H} \end{array}$ <p><i>...Propanoic acid...</i></p>	
$\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{C} \\ \quad // \quad \backslash \\ \text{H} \quad \text{O} \quad \text{OH} \end{array}$ <p><i>...Ethanoic acid...</i></p>	
$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \\ \quad \quad \\ \text{H}-\text{C}-\text{C}-\text{C}-\text{C}=\text{O} \\ \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{O}-\text{H} \end{array}$ <p><i>...Butanoic acid...</i></p>	

4. Sarah and Dean carry out an experiment with ethanoic acid solution.

They add the substances in the left-hand boxes, and test any gas formed.

Draw straight lines to link together the substance added with the correct observation.

Substance being added

Observation

Magnesium	Gas formed that turns limewater milky
Sodium hydroxide solution – an alkali	Gas formed that gives a squeaky pop
Chalk – calcium carbonate	No gas formed