Electrolysis using a microscale Hoffman apparatus – student sheet

In this experiment you will be investigating the electrolysis of sodium sulfate solution using a microscale Hoffman apparatus.

Instructions

- 1. Set up the Hoffman apparatus in a clamp.
- 2. Pour approximately 40 cm³ of the sodium sulfate solution into a beaker and add a few drops of bromothymol blue indicator. Note the colour of the solution.
- 3. Using a pipette carefully fill the electrolysis apparatus with the sodium sulfate solution.
- 4. Plug the tops of each stem with a small piece of Blu-Tack®.
- 5. Carefully attach the crocodile clips to the electrodes and record all your observations over the next 15 minutes.
- 6. Disconnect the leads and try to give explanations for your observations.

Question

Can you think of a way of testing for either of the gases that you have collected?

Health, safety and technical notes

- 1. Wear eye protection throughout.
- 2. Check the information sheet on microscale apparatus to find out more about setting up the Hoffman apparatus.
- 3. Sodium sulfate, $Na_2SO_3(aq)$, 0.5 mol dm⁻³ is of low hazard.
- 4. Bromothymol blue solution is of low hazard.

