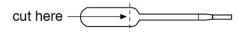
Unsaturation test with potassium manganate(VII) - student sheet

In this experiment, you will use a solution of potassium manganate(VII) in propanone to detect whether an organic compound is unsaturated.

Instructions

1. Cut the end off a plastic pipette as shown below and place the cup in a glass beaker or test tube.



- 2. Carefully add a few crystals of potassium manganate(VII) to the cup.
- 3. Add propanone to the cup until it is about half-full. You will notice that the potassium manganate(VII) dissolves to give a purple solution. Is it surprising that potassium manganate(VII) dissolves in an organic solvent?
- 4. Cut the ends off three pipettes to make small reaction vessels as shown below and place them in the lid of a plastic Petri dish.



- 5. Using a plastic pipette, add four drops of the potassium manganate(VII) in propanone solution to each of the reaction vessels.
- 6. Put three drops of each of the organic liquids under test in the reaction vessel and observe any changes over the next few minutes.
- 7. Mop up the liquid with tissue paper when you have finished.

Question

Which types of organic liquids react with potassium manganate(VII)?

Health, safety and technical notes

- Wear eye protection throughout.
- Avoid sources of ignition.
- Propanone is HIGHLY FLAMMABLE.
- Potassium manganate(VII) is an OXIDISER and IRRITANT, and stains glass, plastic, clothing and skin. Avoid direct contact and store in the dark.
- Cyclohexane is HIGHLY FLAMMABLE, a skin/respiratory IRRITANT and very TOXIC to aquatic life.
- Cyclohexene is a HIGHLY FLAMMABLE liquid and vapour. It is HARMFUL if ingested or in contact with the skin and is very TOXIC to aquatic life.
- Limonene is FLAMMABLE as liquid and vapour. It is TOXIC to aquatic life and is an IRRITANT and SENSITISER to skin.

