Bonding workshop

Covalent bonding and hydrocarbons

Download the teacher notes, PowerPoint presentation and student workbook that accompany this resource at <u>rsc.li/3ASiknU.</u>

Read our health and safety guidance, available from <u>rsc.li/3IAmFA0</u>, and carry out a risk assessment before running any live practical. The safety equipment suggested is in line with CLEAPSS requirements. For non-hazardous substances, wearing lab coats can help protect clothes. The safety rules might be different where you live so it is worth checking local and school guidance.

Acknowledgements

This resource was originally developed by Liverpool John Moores University to support outreach work delivered as part of the Chemistry for All project.

To find out more about the project, and get more resources to help widen participation, visit our Outreach resources hub: <u>rsc.li/3CJX7M3.</u>



Activity 1: hydrocarbons

No equipment required.

Activity 2: structure of hydrocarbons

Apparatus for room:

- Computer and projector
- Whiteboards and pens

Activity 3: simple molecules

Molymod kit, available from: <u>rsc.li/3GT6oVD</u>

Activity 4: alkanes

Equipment per pair of learners:

- Molymod kits, available from: <u>rsc.li/3GT6oVD</u>
 - Each kit needs to contain at least:
 - 2 × Cl, 19 × H, 2 × O, 8 × C, 1 × N
 - 30 single bonds
 - 4 double bonds

Activity 5: reactions of alkanes

One set of equipment for the facilitator needed per demonstration.

Demonstration 1: Bunsen burner flame

- Bunsen burner
- Heatproof mat
- Matches/lighter



Demonstration 2: controlled explosion of methane



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Read the safety notes and method for the Exploding a tin can using methane experiment, available from <u>rsc.li/3vPcMqv</u>, before demonstrating it.

Equipment

- Eye protection for the demonstrator
- Ear defenders
- Two safety screens
- A large tin with a press-on lid (A catering size (500–750 g) instant coffee tin works well.)
- Glass tubing, 2-3 cm in diameter, 30-50 cm long
- Epoxy resin adhesive (eg Araldite)
- Length of rubber tubing
- Boss, clamp and stand
- Wooden splints

Chemicals

• Methane (natural gas) supply (DANGER: extremely flammable)



Demonstration 3: methane bubbles

Read the safety notes and method for the Exploding soap bubbles experiment, available from <u>rsc.li/3GqsNls</u>, before demonstrating it.

Equipment:

- Oxygen cylinder
- Methane from gas tap
- Rubber tubing
- Metre rule
- Wooden splints
- Tape
- Plastic washing up bowl
- Washing up liquid
- Ear defenders
- Eye protection
- Warm water



Activity 6: alkenes

Equipment per pair of learners:

- Molymod kits
- Each kit needs to contain at least:
 - 2 × Cl, 19 × H, 2 × O, 8 × C, 1 × N
 - 30 single bonds
 - 4 double bonds

Activity 7: reactions of alkenes

Class practical/demonstration: decolourisation of bromine water



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Required per group

Two stoppered test tubes containing 0.5 cm^3 of $0.02 \text{ mol } dm^{-3}$ bromine water in a test tube rack. (See CLEAPSS method for making bromine water, available from <u>bit.ly/3vMyBqL</u>.)

Required per three/four groups

Two dropping bottles – one containing hexane and one containing hex-1-ene.

Health and safety

Provide eye protection for each learner. The safety equipment suggested is in line with CLEAPSS requirements. For non-hazardous substances, wearing lab coats can help protect clothes. The safety rules might be different where you live so it is worth checking local and school guidance.



Please also see CLEAPSS Hazcards:

- 15A Bromine liquid, available from <u>bit.ly/3Cz0gze</u> (See also 15B: <u>bit.ly/3vQ3VVw</u>)
- 45B Hydrocarbons: aliphatic, saturated, available from <u>bit.ly/3H0nlxv</u>
- 45C Hydrocarbons: aliphatic & unsaturated, available from <u>bit.ly/3vQ5h2y</u>

Useful CLEAPSS protocols:

TL010 – unsaturation tests on hydrocarbons, available from <u>bit.ly/3k5eEIT</u>

Disposal: Place the test tubes, and contents, into a bowl of water in the fume cupboard.

Activity 8: functional groups

Equipment per pair of learners:

- Molymod kits
- Each kit needs to contain at least:
 - 2 × Cl, 19 × H, 2 × O, 8 × C, 1 × N
 - 30 single bonds
 - 4 double bonds

