



Chromatography: knowledge check

1.1 Learners are completing an experiment to separate the coloured soluble substances in two different food dyes.

The diagram shows the apparatus used and the chromatogram produced.

Label the diagram using some of the words provided.

chromatography paper

origin line

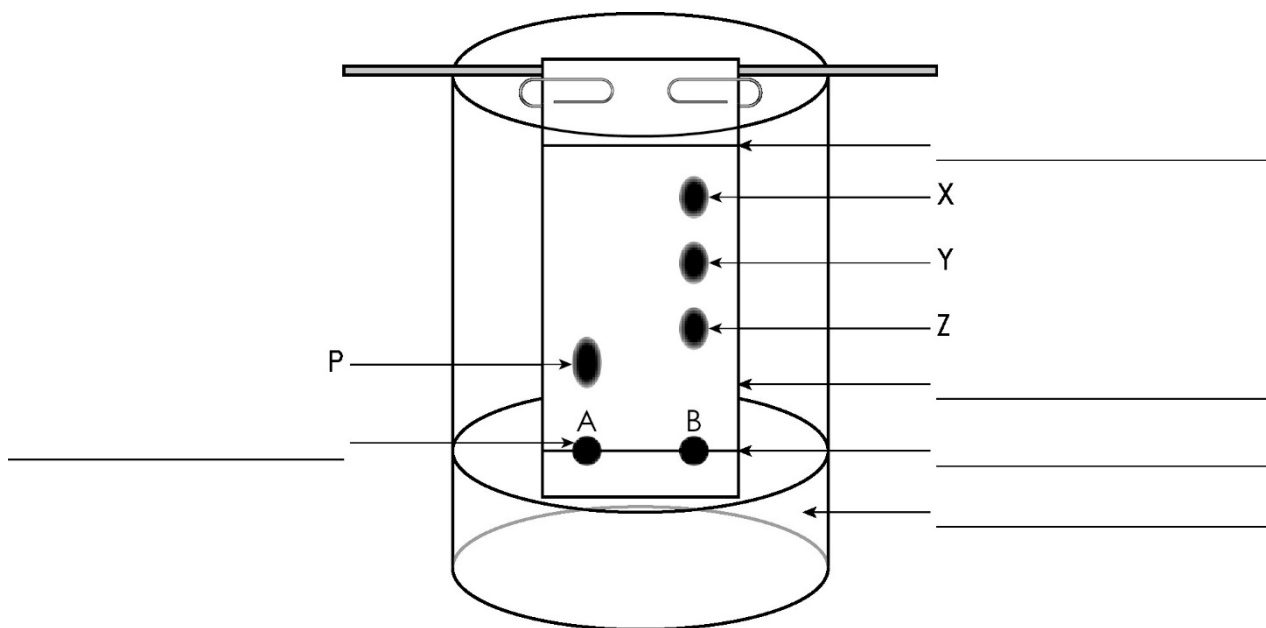
solvent

solution

solvent front

food colouring

solute



Questions 1.2 to 1.4 are about the experiment in question 1.1.



1.2 The table shows the steps in the chromatography process used to separate the coloured substances, but they are given in the wrong order.

Add numbers to show the correct order for the steps. The first has been done for you.

1	Draw an origin line in pencil about 1.5 cm from one end of the chromatography paper.
	Place the paper inside the beaker. Make sure it just touches the water and it is vertical.
	Check the paper is the right length by lining it up on the outside of the beaker so that the water is below the origin line.
	Use a pipette to add small drops of food colourings A and B on the chromatography paper.
	Allow the solvent to move through the paper, removing it before the solvent reaches the top.

1.3 Use some of the words to complete the gaps in the following sentences.

rates mobile phase stationary phase

chromatography paper solvent solute solution

The chromatography paper is the _____ . The solvent is the _____ . The different dissolved substances in a mixture are attracted to the _____ and the chromatography paper in different proportions. This causes them to move at different _____ up the _____ .



1.4 Use some of the words to complete the gaps in the following sentences.

pure **impure** **compound** **one**

two or more **three**

A pure substance contains one type of element or _____ only.

An impure substance contains _____ different elements or compounds. Food colouring A consists of one substance and is a _____ substance.

Food colouring B contains _____ different coloured substances.

Food colouring B is an _____ substance.

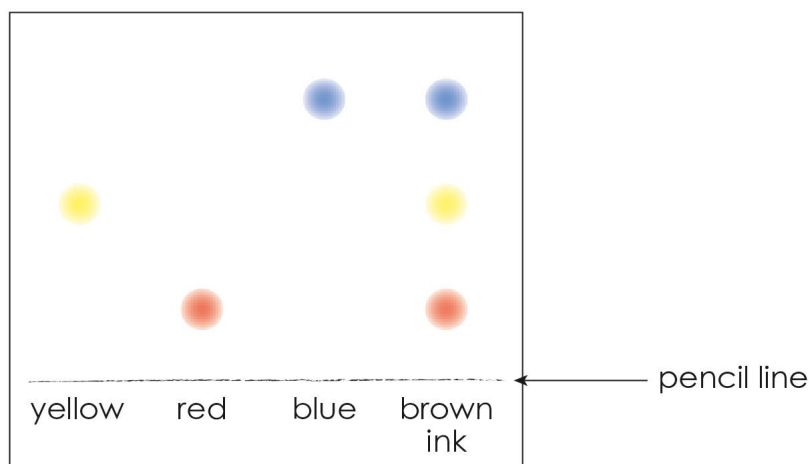


Chromatography: test myself

2.1 What type of mixtures can paper chromatography be used to separate?

[Hint: Think about whether the substances would be soluble or insoluble.]

2.2 This is a chromatogram of four different inks.



Circle the inks on the chromatogram that are pure substances.

2.3 How is chromatography used to identify unknown substances?

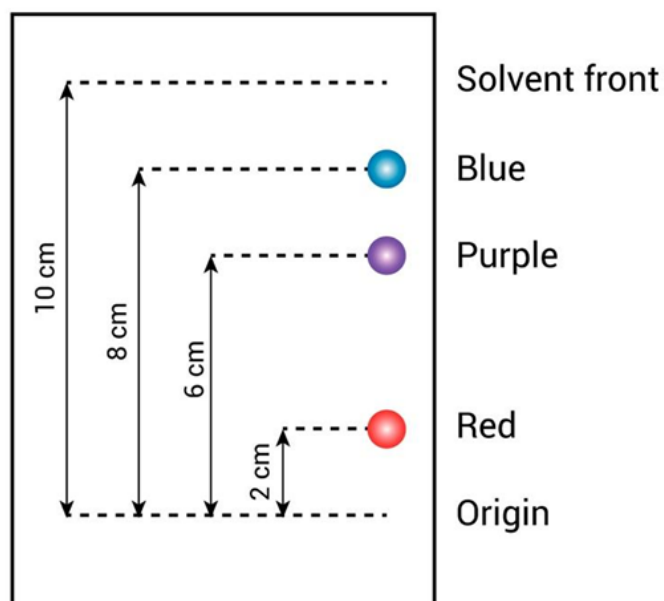
[Hint: Think about what can be measured and compared on a chromatogram.]

2.4 Thin layer chromatography uses a glass plate coated with alumina or silica gel.

What does this glass plate replace in the experiment in **question 1.1**?



2.5 This is a chromatogram of dark blue ink.



The equation used to calculate the R_f value is:

$$R_f = \frac{\text{distance travelled by substance}}{\text{distance travelled by solvent}}$$

Use the data shown in the chromatogram, along with the equation, to calculate the correct R_f values for the red, purple and blue substances.

Substance	R_f values
Red	
Purple	
Blue	



2.6 A compound has a R_f value of 0.6. If the solvent travels 16.0 cm up the chromatography paper, what is the distance travelled by the compound?

Use the equation to calculate the answer in cm.

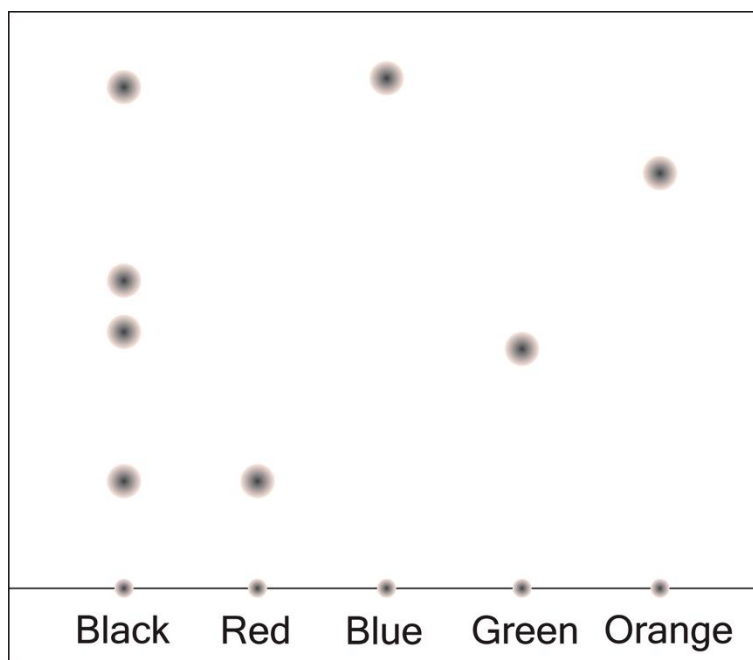
distance travelled by substance = $R_f \times$ distance travelled by solvent



Chromatography: feeling confident?

3.1 This is a chromatogram of different coloured inks.

What can you conclude about the black ink from the chromatogram shown?



Use the words in your answer. The first sentence has been started to help you.

black ink **red** **blue** **green** **orange**

My conclusion is that the **black ink** contains



3.2 Substances **A**, **B** and **C** are found in chlorophyll. The R_f values of these three substances can be determined using thin layer chromatography and an organic solvent.

The equation used to calculate the R_f value is:

$$R_f = \frac{\text{distance travelled by substance}}{\text{distance travelled by solvent}}$$

Use the equation to calculate the missing values in the table.

Substance	Distance travelled by the substance when the solvent travels 10 cm (cm)	R_f value
A	9.8	
B	5.9	
C		0.42



Chromatography: what do I understand?

Think about your answers and confidence level for each mini-topic. Decide whether you understand it well, are unsure or need more help. Tick the appropriate column.

Mini-topic	I understand this well	I think I understand this	I need more help
I can describe the process of paper chromatography.			
I can describe how to prepare a chromatogram.			
I can define a pure substance and an impure substance			
I can interpret a chromatogram.			
I can calculate R_f values.			
I can use R_f values.			
Feeling confident? topics	I understand this well	I think I understand this	I need more help
I can interpret chromatograms of coloured inks.			
I can interpret and use information from chromatograms of chlorophyll.			