Microscale titration

Integrated instructions

ROYAL SOCIETY OF CHEMISTRY

Education
Inspiring your teaching and learning

Integrated instructions

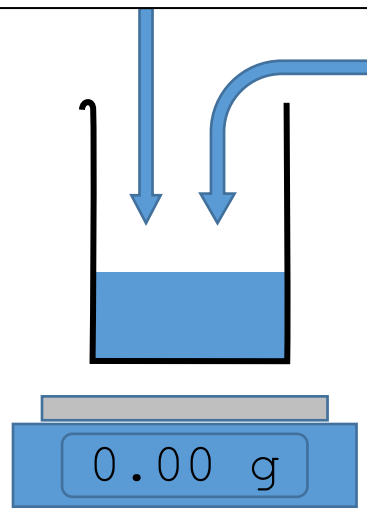
Integrated instructions use clear numbering, arrows and simple pictograms (for example, an eye showing when to make observations). They were developed using cognitive load theory and remove unnecessary information. Integrated instructions reduce extraneous load on learners, increasing the capacity of their working memory to think about what they are doing and why.

Read more at rsc.li/47bIKi5.

Full technician notes, including safety, preparation and disposal, are available from rsc.li/4iclogx.

❶ 1 drop phenolphthalein. Take the mass, **M1** ❑

❷ 1 cm^3 vinegar. Take the mass, **M2** ❑



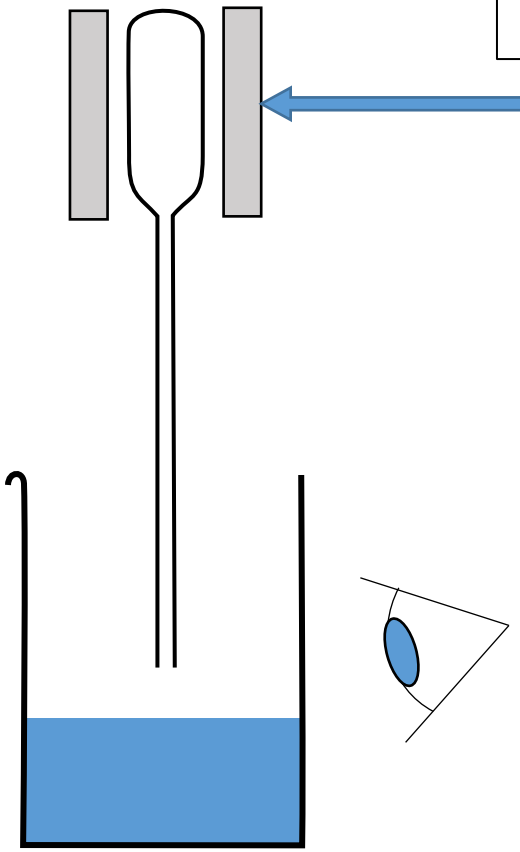
❸ Fill pipette with NaOH ❑

❹ Clamp pipette gently ❑

❺ Carefully tighten clamp to add 1 drop NaOH ❑

❻ Swirl vial ❑

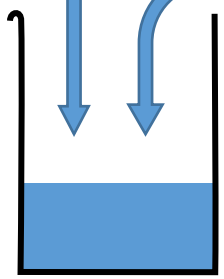
❼ Repeat steps **5** and **6** until permanent pink. Take the mass, **M3** ❑



RSC Microscale chemistry
Gravimetric titration
Downloaded from rsc.li/4iclogx
Based on a method developed by CLEAPSS

❶ 1 drop phenolphthalein ❑

❷ 1 cm^3 vinegar ❑



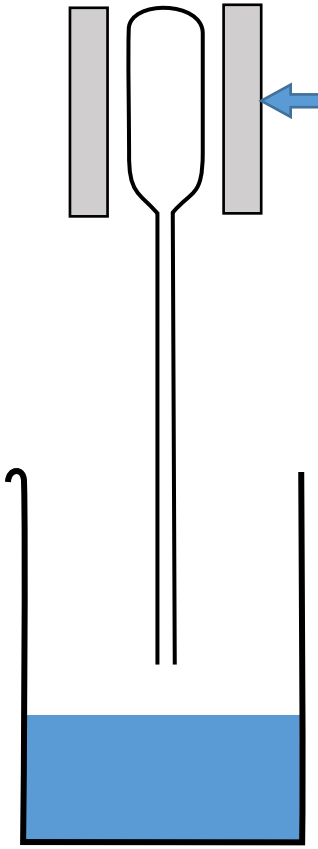
❸ Fill pipette with $NaOH$ ❑

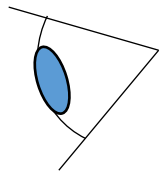
❹ Clamp pipette gently ❑

❺ Carefully tighten clamp to add 1 drop $NaOH$ ❑

❻ Swirl vial ❑

❼ Repeat steps **5** and **6**, counting the drops, until permanent pink ❑





RSC Microscale chemistry
Volumetric titration
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