

Scientists create liquid carbon on Earth for the first time

Original article by Victoria Corless. Adapted by Nina Notman.

Researchers reveal breakthrough information about carbon's most mysterious form using an ultra-powerful laser

Researchers have successfully characterised liquid carbon in the lab using an ultra-powerful laser. The data obtained offers rare insight into one of the most abundant elements in the universe which, despite its ubiquity, remains the least understood of the stable elements in its liquid form.

Unknown properties

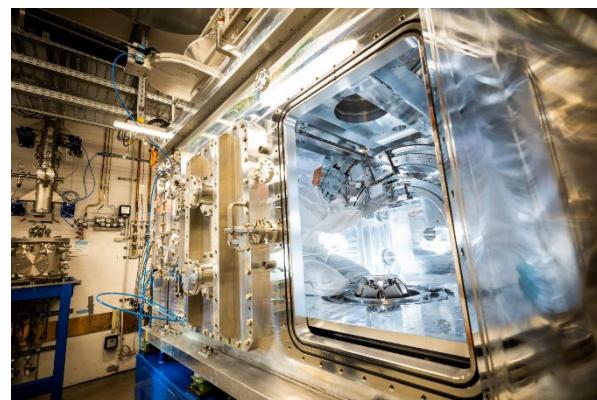
Liquid carbon is notoriously difficult to make in the lab because it only exists under extreme temperatures and pressures. 'The liquid phase requires pressures of at least several hundred atmospheres to form,' says Dominik Kraus, a physicist from the University of Rostock in Germany, who led the study. Carbon also has the highest melting temperature of all materials under these conditions, meaning there is no known material that can contain liquid carbon while you analyse it in the lab.

Without the ability to make liquid carbon in the lab, scientists are unable to study it, and its properties remain largely unknown. Liquid carbon forms transiently during the synthesis of some novel carbon materials, such as [carbon nanotubes](#) and nanodiamonds, and during nuclear fusion. Understanding carbon's behaviour in liquid form will help scientists better understand these processes.

Scientists also believe that liquid carbon exists deep within the interiors of icy giant planets, such as Uranus and Neptune. 'Studying liquid carbon [will help] model how carbon behaves deep within these planets,' says Dominik. 'It may undergo unusual changes that could affect a planet's heat flow, magnetic field and overall structure.'

Going to extremes

To create liquid carbon on Earth, Dominik and the team used a record-breaking solid-state laser that was developed by a team at the Centre for Advanced Laser Technology and Applications (CALTA) in the UK. The researchers used the laser to create pressures exceeding one million atmospheres and temperatures of around



Source: © Jan Hosan/European XFEL

The experiments used the High Energy Density instrument at the European XFEL facility near Hamburg, Germany

7000 K. This turned solid carbon into liquid form for a few nanoseconds – long enough to take snapshots of the material using x-ray.

This is by far 'the most detailed and informative data yet obtained for the liquid state of carbon,' comments physical chemist Richard Saykally from the University of California, Berkeley, in the US, who was not part of the project team. The x-ray diffraction data closely aligns with predictions, he adds.

In [solid diamond](#), each carbon atom is bonded to four others in a stable, fixed structure. When melted, researchers expected these bonds to break completely. But Dominik and his team found that the atoms in liquid carbon tended to stay near four neighbours. 'It somewhat preserves remnants from the solid diamond structure,' says Dominik.

The team now plans to expand its laser-based techniques to observe other understudied materials under extreme pressure and temperature.

This is adapted from the article **Liquid carbon characterised in the lab for the first time** in *Chemistry World*. Read the full article: bit.ly/44UgrEA