**Alcohols: knowledge check**

1. The table shows four different representations of a molecule of the alcohol ethanol.

Use the words provided to identify each of the representations and circle the alcohol functional group shown in each image.

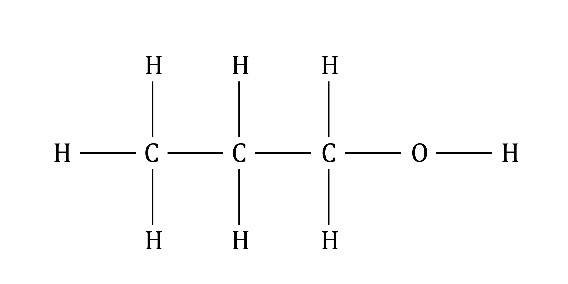
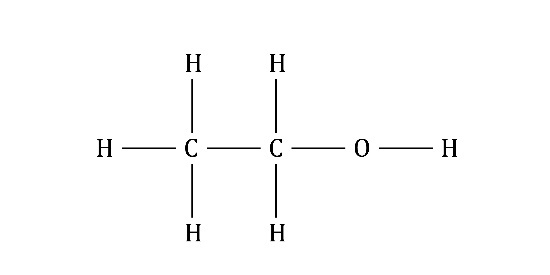
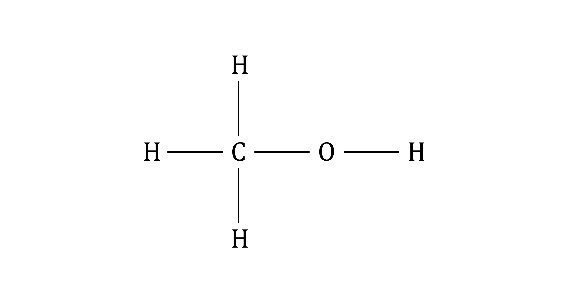
**displayed structural formula molecular formula**

**condensed structural formula ball and stick model**

|  |  |  |  |
| --- | --- | --- | --- |
| **A**  There is a photograph of a model of a molecule on a grey background. From left to right, there is a large black sphere connected to three smaller white spheres and also attached to another large black sphere. That black sphere is attached to two smaller white spheres and one medium sized red sphere. The red sphere is connected to one small white sphere. | **B**  **There is a diagram showing two letters C in the middle joined by a single line. The left hand C has three letters H joined to it by single lines. The right hand C has two letters H joined to it by single lines and a letter O joined to it by a single line. The letter O then has a letter H joined to it by a single line.** | **C**  CH3CH2OH | **D**  C2H5OH |
|  |  |  |  |

1. The table on page 2 includes the names, molecular formulas, condensed structural formulas and displayed structural formulas for the first four members of the alcohols homologous series.

Use some of the formulas provided to complete the table.

**CH3CH2OH C4H9OH CH3CH2CH2CH2OH**

**CH3OH C2H5OH CH3CH2CH2OH**

|  |  |  |  |
| --- | --- | --- | --- |
| **Name** | **Molecular formula** | **Condensed structural formula** | **Displayed structural formula** |
| methanol |  | CH3OH |  |
| ethanol |  |  |  |
| propan-1-ol | C3H7OH |  |  |
| butan-1- ol |  |  | There is a diagram showing four letters C in the middle joined by  single lines. The left hand C has three letters H joined to it by single lines. The two middle Cs have two letters H joined to them by single lines and the right hand C has two letters H joined by single lines and a letter O joined to it by a single line. The letter O then has a letter H joined to it by a single line. |

1. Use some of the words provided to complete the gaps in the following sentences. Not all of the words will be used.

**–ol physical CnHnOH –al –OH CnH2n+1OH**

**chemical letters C=O numbers**

Alcohols are a homologous series with the functional group \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ .

The general formula of alcohols is \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ .

All alcohols have names ending in \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ .

Alcohols with three or more carbon atoms in their molecules have \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ in their names to show the position of the –OH group.

The \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ properties, such as boiling points, vary gradually as the molecules increase in size.

1. Select the correct number or words from those provided in the list to complete the sentences and equations. Not all of the numbers or terms will be used. Some might be used twice.

**ethanoic acid propanoic acid hydrogen**

**2 3 addition physical combustion chemical**

**oxidised oxygen reduced CH3CH2OH**

All alcohols have similar \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ properties.

Alcohols reacts with oxygen in \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ reactions. The equation for the complete combustion of ethanol is:

CH3CH2OH + \_\_\_\_\_\_\_\_\_O2 → \_\_\_\_\_\_\_\_\_CO2 + \_\_\_\_\_\_\_\_\_H2O

If a small piece of sodium is dropped into an alcohol, it reacts steadily and gives off \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ gas. A solution of sodium ethoxide is produced.

Alcohols can also be \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ to form carboxylic acids using an oxidising agent such as potassium manganate(VII).

The equation for the oxidation of ethanol is:

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_. + 2[O] → CH3COOH + H2O

The carboxylic acid produced is named \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

**Alcohols: test myself**

1. Which **two** compounds are alcohols?

**ethanoic acid decan-1-ol propane ethanal**

**propanone propan-2-ol propanal**

1. A molecule of pentan-1-ol has five carbon atoms. What is its molecular formula?
2. Name the carboxylic acid produced when butan-1-ol is oxidised.

[*Hint: Think about the name of the alcohol that ethanoic acid is produced from.*]

1. Complete the equation representing the oxidation of propan-1-ol.

[*Hint: Look back at* ***question******1.4*** *to remind yourself about the products of alcohol oxidation.*]

CH3CH2CH2OH + 2[O] → \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. Use some of the numbers and formulas provided to complete the balanced symbol equation for the complete combustion of propan-1-ol.

**2 3 6 8 9**

**C CO2 H2O O2 CO**

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_C3H7OH + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ → \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. Ethanol can be produced industrially from ethene gas. What is ethene gas reacted with to produce ethanol in this process?

[*Hint: Think about the difference between the functional groups of ethene and ethanol.*]

1. Ethanol can also be produced from biomass such as sugar beet.
   1. What is the name of the process used?

[*Hint: Think about how the process used to make alcoholic drinks such as wine or beer.*]

* 1. Statethe three conditions required to produce ethanol from biomass.

[*Hint: Think about the temperature needed and whether any water or a catalyst are needed.*]

1. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
2. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
3. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
4. Ethanol can be separated from a mixture of ethanol and water.
   1. What is the name of the process used to separate the ethanol?
   2. What property does the separation of ethanol and water depend on when using the method identified in (a)?

**Alcohols: feeling confident?**

1. The image shows the apparatus used by learners to prepare a solution of ethanol.

There is one boiling tube and one test tube connected by a thin tube. The boiling tube on the left is three-quarters full of a yellowish liquid. On top of that liquid is a thin pale brown layer. The tube is stoppered, with a thin tube starting above the liquid layers, going through the stopper. The tube leads to an open test tube on the right. The test tube is about half full of a pale blue liquid. The open end of the tube finishes under the blue liquid.

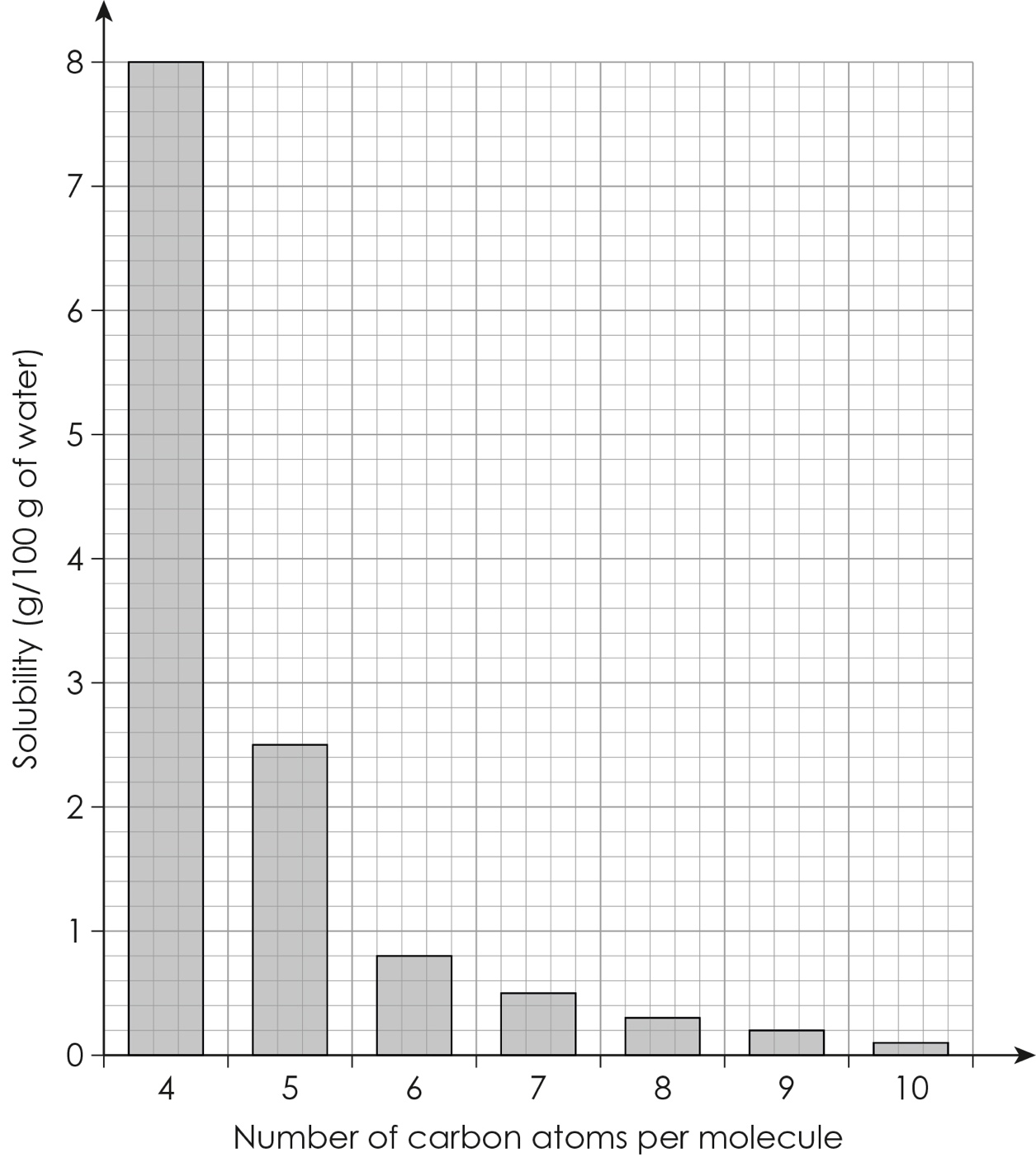
Label the diagram and complete the sentences to describe this process.

The \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ in yeast act as a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ and convert glucose into ethanol and \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ . The gas produced bubbles through the limewater and turns it \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ . The equation for the reaction is:

C6H12O6 → 2C2H5OH + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. The graph shows how the solubility of alcohols in water changes as the number of carbon atoms in the molecules increase.

Use the graph to answer the following questions.



* 1. Which is more soluble in water, butan-1-ol or hexan-1-ol?

[*Hint: Think about what the name of the alcohol tells you about the number of carbon atoms in the molecule.*]

* 1. How many grams of pentan-1-ol dissolve in 100 g water?
  2. What conclusion can be made about the relationship between the number of carbon atoms per molecule and the solubility of the alcohol?
  3. What would you expect to see when decan-1-ol is mixed with water? Explain your answer.

[*Hint: Think about the solubility of decan-1-ol.*]

I would expect to see \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ because

**Alcohols: what do I understand?**

Think about your answers and confidence level for each mini-topic. Decide whether you understand it well, are unsure or need more help. Tick the appropriate column.

|  |  |  |  |
| --- | --- | --- | --- |
| **Mini-topic** | **I understand  this well** | **I think I understand this** | **I need more  help** |
| I know that alcohols are a homologous series. |  |  |  |
| I can identify the functional group and the molecular and general formulae of alcohols. |  |  |  |
| I can draw the condensed structural and displayed structural formulas of alcohols. |  |  |  |
| I understand how alcohols are named. |  |  |  |
| I can describe the combustion reactions of alcohols. |  |  |  |
| I can describe the reactions of alcohols with sodium metal and with oxidising agents. |  |  |  |
| I can describe how ethanol is produced industrially. |  |  |  |
| I can describe the process of fermentation. |  |  |  |
| **Feeling confident? topics** | **I understand  this well** | **I think I understand this** | **I need more  help** |
| I can describe a fermentation reaction and complete the equation. |  |  |  |
| I can describe the solubility of alcohols. |  |  |  |