What happens to particles in chemical reactions?

Name ........................................

When we mixed together *solid* potassium iodide and *solid* lead nitrate

When we added *solutions* of potassium iodide and lead nitrate at the *edge* of the Petri dish

When we added *solutions* of potassium iodide and lead nitrate to the *centre* of the Petri dish

My explanation of what happened when we mixed *solutions* of potassium iodide and lead nitrate is
<table>
<thead>
<tr>
<th>Our apparatus when we set it up:</th>
<th>Our apparatus when we finished:</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
</tr>
</tbody>
</table>

The measurements we made are

What I can tell from our experiment is