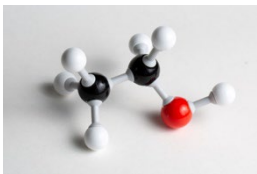




## Alcohols: knowledge check

1.1 The table shows four different representations of a molecule of the alcohol ethanol.

Add the correct name to identify each of the representations and circle the alcohol functional group shown in each.

A	B	C	D
	$  \begin{array}{c}  \text{H} \quad \text{H} \\    \quad   \\  \text{H}-\text{C}-\text{C}-\text{O}-\text{H} \\    \quad   \\  \text{H} \quad \text{H}  \end{array}  $	$\text{CH}_3\text{CH}_2\text{OH}$	$\text{C}_2\text{H}_5\text{OH}$

1.2 The table below includes the names, molecular formulas, condensed structural formulas and displayed structural formulas for the first four members of the alcohols homologous series. Complete the table by filling in the missing formulas.

Name	Molecular formula	Condensed structural formula	Displayed structural formula
methanol			
ethanol			
propan-1-ol			
butan-1-ol			



**1.3** Add the correct terms to complete the gaps in the following sentences.

Alcohols are a homologous series with the functional group \_\_\_\_\_.

The general formula of alcohols is \_\_\_\_\_.

All alcohols have names ending in \_\_\_\_\_.

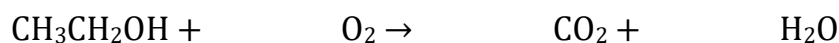
Alcohols with three or more carbon atoms in their molecules have \_\_\_\_\_ in their names to show the position of the -OH group.

The \_\_\_\_\_ properties, such as boiling points, vary gradually as the molecules increase in size.

**1.4** Add the correct numbers or terms to complete the sentences.

All alcohols have similar \_\_\_\_\_ properties.

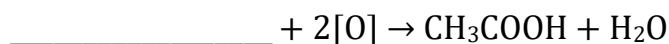
Alcohols reacts with oxygen in \_\_\_\_\_ reactions. The equation for the complete combustion of ethanol is:



If a small piece of sodium is dropped into an alcohol, it reacts steadily and gives off \_\_\_\_\_ gas. A solution of sodium ethoxide is produced.

Alcohols can also be \_\_\_\_\_ to form carboxylic acids using an oxidising agent such as potassium manganate(VII).

The equation for the oxidation of ethanol is:



The carboxylic acid produced is \_\_\_\_\_.



## Alcohols: test myself

2.1 Which **two** compounds are alcohols?

ethanoic acid    decan-1-ol    propane    ethanal  
propanone    propan-2-ol    propanal

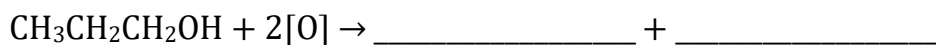
2.2 What is the molecular formula for a molecule of pentan-1-ol?

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2.3 Name the carboxylic acid produced when butan-1-ol is oxidised.

---

2.4 Complete the equation representing the oxidation of propan-1-ol.



2.5 Write the balanced symbol equation representing the complete combustion of propan-1-ol.



2.6 Ethanol can be produced industrially from ethene gas. What is ethene reacted with to produce ethanol?

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**2.7** Ethanol can also be produced from biomass such as sugar beet.

(a) What is the name of the process used?

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(b) State the three conditions required to produce ethanol from biomass.

1 

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2 

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3 

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**2.8** Ethanol can be separated from a mixture of ethanol and water.

(c) What is the name of the process used to separate the ethanol?

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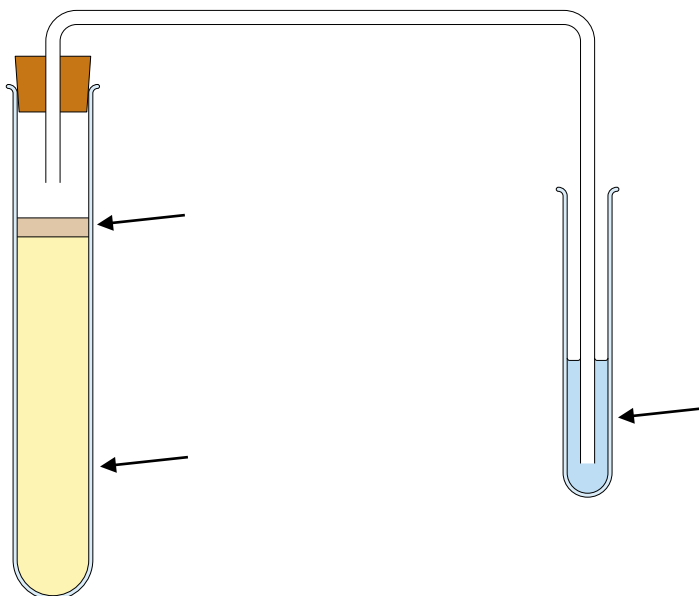
(d) What property does the separation of ethanol and water depend on when using the method identified in (c)?

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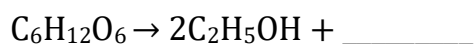
## Alcohols: feeling confident?

3.1 The image shows the apparatus used by learners to prepare a solution of ethanol.



Label the diagram and complete the sentences to describe this process.

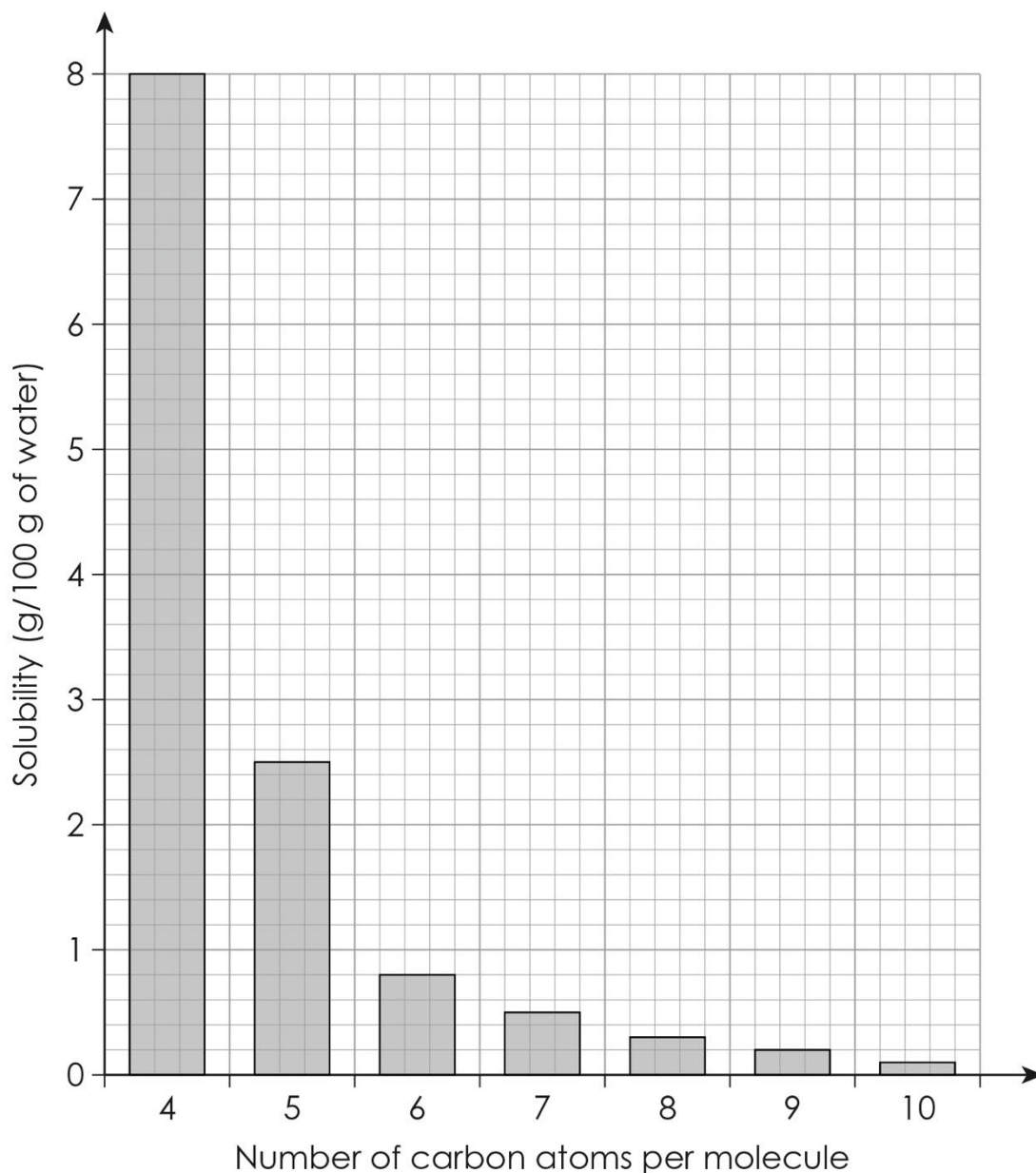
The \_\_\_\_\_ in yeast act as a \_\_\_\_\_ and convert glucose into ethanol and \_\_\_\_\_. The gas produced bubbles through the limewater and turns it \_\_\_\_\_. The equation for the reaction is:





**3.2** The graph shows how the solubility of alcohols in water changes as the number of carbon atoms in the molecules increase.

Use the graph to answer the following questions.



(a) Which is more soluble in water, butan-1-ol or hexan-1-ol?

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(b) How many grams of pentan-1-ol dissolve in 100 g water?

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- (c) What conclusion can be made about the relationship between the number of carbon atoms per molecule and the solubility of the alcohol?

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- (d) What would you expect to see when decan-1-ol is mixed with water?  
Explain your answer.

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## Alcohols: what do I understand?

Think about your answers and confidence level for each mini-topic. Decide whether you understand it well, are unsure or need more help. Tick the appropriate column.

Mini-topic	I understand this well	I think I understand this	I need more help
I know that alcohols are a homologous series.			
I can identify the functional group and the molecular and general formulae of alcohols group of alcohols.			
I can draw the condensed structural and displayed structural formulas of alcohols.			
I understand how alcohols are named.			
I can describe the combustion reactions of alcohols.			
I can describe the reactions of alcohols with sodium metal and with oxidising agents.			
I can describe how ethanol is produced industrially.			
I can describe the process of fermentation.			
Feeling confident? topics	I understand this well	I think I understand this	I need more help
I can answer extending questions on fermentation.			
I can describe the solubility of alcohols.			