Education in Chemistry 16–18 years

Amino acids, peptides and proteins

This resource accompanies the article **A meaty problem** in Education in Chemistry which can be viewed at: <u>https://rsc.li/3IRKgA7</u>

Learning objectives

- 1 Review the naming and structures of common amino acids.
- 2 Draw the structural formulas for dipeptides and tripeptides formed during protein synthesis and the component amino acids formed during hydrolysis.
- 3 Investigate how thin layer chromatography (TLC) can be used to separate amino acids in a mixture.
- 4 Calculate Rf values to identify individual amino acids.

Introduction

Students successfully completing the questions will show understanding of the learning objectives.

'Review the naming and structure of common amino acids' is assessed using questions 1(a), 3 and 5 (b).

'Draw the structural formulas for dipeptides and tripeptides formed during protein synthesis' is assessed using question 2. 'Draw the structural formulas for... the component amino acids formed during hydrolysis' is assessed using questions 3 and 5.

'Investigate how thin layer chromatography (TLC) can be used to separate amino acids in a mixture' is assessed using questions 4 and 5.

'Calculate Rf values to identify individual amino acids' is assessed using questions 4 and 5 (f).



Answers

(b)

1. (a) Glycine: 2-amino ethanoic acid Alanine: 2-amino propanoic acid

Common properties	Reason
They are solids of high melting temperatures	Owing to strong electrostatic attraction between ions
Soluble in water	Strong electrostatic attraction between ions and polar water molecules

2. (a) Gly-Ala:





(b) Gly-Ser-Ser:



Ser-Gly-Ser:





(c) 2 moles of H₂O are formed

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3. Equation should show:

- Addition of 2 moles of water to 1 mole of glutathione as reactants and
- Formulae of constituent amino acids glutamic acid, cysteine and glycine as products.



4.

Amino acid	Estimated Rf value	Corrected Rf value
Alanine (ala)	0.74	2.8/3.5 = 0.80*
Glutamic acid (glu)	0.60	2.2/3.5 = 0.63*
Glycine (gly)	0.46	1.6/3.5 = 0.46*
Serine (ser)	0.22	0.8/3.5 = 0.23*

*Due to printer scaling these values may be different. Please check and amend your mark scheme accordingly.

TEACHER NOTES

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(b)



(c) Serine = most soluble

Both tyrosine and serine contain an O-H bond within the R group which can H-bond with water

In the serine the polar O-H bond is a larger proportion of the R group than in tyrosine

(d)

Amino acid	Side chain polarity	Rf value
Tyrosine	polar	0.59
Serine	polar	0.65
Leucine	non-polar	0.43

(e) Cross contamination using same pipette, Spot too concentrated causing streaking

(f) i. Rf values are 0.8/3.8 = 0.21* and 1.75/3.8 = 0.46*

ii. Tripeptide is made up of only two different amino acids *Due to printer scaling these values may be different. Please check and amend your mark scheme accordingly.