## gridlocks - can you unlock the grid?

## Mass concentration of solutions

Before you answer the gridlocks below fill in the table of concentrations in $\mathrm{g} / \mathrm{dm}^{3}$ using:

$$
\text { mass concentration }=\frac{\text { mass }}{\text { volume }\left(\mathrm{indm}^{3}\right)} \quad \text { and } \quad \text { mass }=\text { moles } \times \text { molar mass }\left(\mathrm{M}_{\mathrm{r}}\right)
$$

| Substance | $\mathbf{M}_{\mathbf{r}}$ | Concentration in <br> $\mathbf{m o l} / \mathbf{d m}^{3}$ | Concentration in <br> $\mathbf{g / d m}$ <br>  <br> $\mathbf{3}$ |
| :---: | :---: | :---: | :---: |
| NaOH | 40 | 0.1 |  |
| HCl | 36.5 |  | 73 |
| $\mathrm{M}_{2} \mathrm{CO}_{3}$ |  |  | 53 |

( $M$ is a metal, you should be able to work out which one)

## Gridlock 1

Each row, column and $2 \times 2$ box contains concentrations of NaOH of $1,0.5,0.1$ and $0.4 \mathrm{~mol} / \mathrm{dm}^{3}$. Use your problem solving skills and the skills you used to put answers in the table above to fill in the blank boxes.

| concentration in $\mathrm{mol} / \mathrm{dm}^{3}$ |  |  | concentration in $\mathrm{g} / \mathrm{dm}^{3}$ |  |
| :---: | :---: | :---: | :---: | :---: |
| 0.5 |  |  |  |  |
|  |  | 16 |  |  |
|  |  | 1 |  |  |
|  |  |  |  |  |

## gridlocks - can you unlock the grid?

## Gridlock 2

Each row, column and $2 \times 2$ box contains information about solutions of NaOH with concentrations of $1,0.5$, 0.1 and $0.4 \mathrm{~mol} / \mathrm{dm}^{3}$.

| concentration in mol/dm ${ }^{3}$ |  | concentration in $\mathrm{g} / \mathrm{dm}^{3}$ |  |
| :---: | :---: | :---: | :---: |
|  | 1 |  |  |
|  |  |  | 16 |
| 0.0125 |  |  | 40 |

## Gridlock 3

Each row, column and $2 \times 2$ box contains information about solutions of NaOH with concentrations of $1,0.5$, 0.1 and $0.4 \mathrm{~mol} / \mathrm{dm}^{3}$.

| concentration in $\mathrm{mol} / \mathrm{dm}^{3}$ |  |  | concentration in $\mathrm{g} / \mathrm{dm}^{3}$ |
| :---: | :---: | :---: | :---: |
| 0.5 |  |  | 40 |
|  |  |  |  |

## gridlocks - can you unlock the grid?

## Mass concentration of solutions - answers

Before you answer the puzzles below fill in the table of concentrations in $\mathrm{g} / \mathrm{dm}^{3} \mathrm{using}$ :

$$
\text { mass concentration }=\frac{\text { mass }}{\text { volume }\left(\mathrm{indm}^{3}\right)} \quad \text { and } \quad \text { mass }=\text { moles } \times \text { molar mass }\left(\mathrm{M}_{\mathrm{r}}\right)
$$

| Substance | $\mathbf{M}_{\mathbf{r}}$ | Concentration in <br> $\mathbf{m o l} / \mathbf{d m}^{\mathbf{3}}$ | Concentration in <br> $\mathbf{g} / \mathbf{d m}^{\mathbf{3}}$ |
| :---: | :---: | :---: | :---: |
| NaOH | 40 | 0.1 | $\mathbf{4}$ |
| HCl | 36.5 | $\mathbf{2}$ | 73 |
| $\mathrm{M}_{2} \mathrm{CO}_{3}$ | $\mathbf{1 0 6}$ | 0.5 | 53 |

( M is sodium)

## Puzzle 1 - answers

Each row, column and $2 \times 2$ box contains concentrations of NaOH of $1,0.5,0.1$ and $0.4 \mathrm{~mol} / \mathrm{dm}^{3}$. Use your problem solving skills and the skills you used to put answers in the table above to fill in the blank boxes.

| concentration in $\mathrm{mol} / \mathrm{dm}^{3}$ |  | concentration in $\mathrm{g} / \mathrm{dm}^{3}$ |  |
| :---: | :---: | :---: | :---: |
| 0.5 | 0.4 | 4 | 40 |
| 0.1 | 1 |  |  |
| 16 | 20 | 16 | 20 |
|  |  | 1 | 0.1 |
| 40 | 4 |  |  |

## grídlOCKS - can you unlock the grid?

Puzzle 2 - answers
Contains information about solutions of NaOH with concentrations of $1,0.5,0.1$ and $0.4 \mathrm{~mol} / \mathrm{dm}^{3}$.

| concentration in mol/dm ${ }^{3}$ |  | concentration in $\mathrm{g} / \mathrm{dm}^{3}$ |  |
| :---: | :---: | :---: | :---: |
| 0.1 | 1 | 20 | 16 |
| 0.4 | 0.5 | 4 | 40 |
| 0.025 | 0.0025 | 0.4 | 0.5 |
| 0.0125 | 0.01 | 1 | 0.1 |
| moles in $25 \mathrm{~cm}^{3}$ |  | concentration in $\mathrm{mol} / \mathrm{dm}^{3}$ |  |

## Puzzle 3 - answers

Contains information about solutions of NaOH with concentrations of $1,0.5,0.1$ and $0.4 \mathrm{~mol} / \mathrm{dm}^{3}$.

| concentration in mol/dm ${ }^{3}$ |  | concentration in $\mathrm{g} / \mathrm{dm}^{3}$ |  |
| :---: | :---: | :---: | :---: |
| 0.5 | 0.1 | 16 | 40 |
| 0.4 | 1 | 20 | 4 |
| 0.025 | 0.0125 | 0.1 | 0.4 |
| 0.0025 | 0.01 | 1 | 0.5 |
| moles in $25 \mathrm{~cm}^{3}$ |  | mass in $25 \mathrm{~cm}^{3}$ in g |  |

