Mass concentration of solutions

Before you answer the gridlocks below fill in the table of concentrations in g/dm³ using:

mass concentration = $\frac{111ass}{\text{volume (in dm}^3)}$

and

mass = moles \times molar mass (M_r)

Substance	Mr	Concentration in mol/dm ³	Concentration in g/dm ³
NaOH	40	0.1	
HCI	36.5		73
M ₂ CO ₃		0.5	53

(M is a metal, you should be able to work out which one)

Gridlock 1

Each row, column and 2 x 2 box contains concentrations of NaOH of 1, 0.5, 0.1 and 0.4 mol/dm³. Use your problem solving skills and the skills you used to put answers in the table above to fill in the blank boxes.

concentration in mol/dm ³		concentration in g/dm ³	
0.5			
		16	
		1	
	4		

concentration in g/dm³

concentration in mol/dm³





Gridlock 2

Each row, column and 2 x 2 box contains information about solutions of NaOH with concentrations of 1, 0.5, 0.1 and 0.4 mol/dm³.

concentration in mol/dm ³		concentration in g/dm ³	
	1		16
			40
0.0125			0.1
moles ir	1 25 cm ³	concentratio	n in mol/dm ³

Gridlock 3

Each row, column and 2 x 2 box contains information about solutions of NaOH with concentrations of 1, 0.5, 0.1 and 0.4 mol/dm³.

concentration in mol/dm ³		concentration in g/dm ³	
0.5			40
			0.4
0.0025			
moles ir	1 25 cm ³	mass in 2	5 cm ³ in a

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Mass concentration of solutions – answers

Before you answer the puzzles below fill in the table of concentrations in g/dm³ using:

mass concentration = $\frac{111ass}{\text{volume (in dm^3)}}$

```
and
```

mass = moles \times molar mass (M_r)

Substance	Mr	Concentration in mol/dm ³	Concentration in g/dm ³
NaOH	40	0.1	4
HCI	36.5	2	73
M ₂ CO ₃	106	0.5	53

(M is sodium)

Puzzle 1 – answers

Each row, column and 2 x 2 box contains concentrations of NaOH of 1, 0.5, 0.1 and 0.4 mol/dm³. Use your problem solving skills and the skills you used to put answers in the table above to fill in the blank boxes.

concentration in mol/dm ³		concentration in g/dm ³	
0.5	0.4	4	40
0.1	1	16	20
16	20	1	0.1
40	4	0.5	0.4
<u> </u>			

concentration in g/dm³

concentration in mol/dm³





Puzzle 2 – answers

Contains information about solutions of NaOH with concentrations of 1, 0.5, 0.1 and 0.4 mol/dm³.

concentration in mol/dm ³		concentration in g/dm ³	
0.1	1	20	16
0.4	0.5	4	40
0.025	0.0025	0.4	0.5
0.0125	0.01	1	0.1
moles in 25 cm ³		concentratio	n in mol/dm ³

moles in 25 cm³

Puzzle 3 – answers

Contains information about solutions of NaOH with concentrations of 1, 0.5, 0.1 and 0.4 mol/dm³.

concentration in mol/dm ³		concentration in g/dm ³	
0.5	0.1	16	40
0.4	1	20	4
0.025	0.0125	0.1	0.4
0.0025	0.01	1	0.5
moles in 25 cm ³		mass in 2	5 cm ³ in g

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